


```
1 matplotlib.rc("image",cmap="gray")
2
3 import urllib2
4 from urllib2 import urlopen
5 from StringIO import StringIO
6 from PIL import Image
7 def fig(url):
8     data = urlopen(url).read()
9     axis("off"); return imshow(Image.open(StringIO(data)))
10
11 from pylab import *
12 from scipy.ndimage import filters,measurements
13 import pygame
14 from pygame import surfarray
15 matplotlib.rc("image",cmap="gray")
16 matplotlib.rc("image",interpolation="nearest")
17 pygame.init()
18 OR = logical_or
19 AND = logical_and
20 def unif(lo,hi,size=1):
21     if size==1: return rand()*(hi-lo)+lo
22     if type(size)==int: return rand(size)*(hi-lo)+lo
23     return rand(*size)*(hi-lo)+lo
24 def animate(images,n=20,k=1):
25     for i,image in enumerate(images):
26         if i>=n: break
27         if i%k!=0: continue
28         b = array(255*clip(image,0,1),'B')
```

Basic Reaction Diffusion Systems

General reaction-diffusion equation:

$$\frac{\partial q}{\partial t} = D \nabla^2 q + R(q)$$

This is like the simple diffusion equation, but with two differences:

- ▶ q may be a vector, so it is multiple signals that diffuse
- ▶ there is an additional term $R(q)$ that represents reactions

The matrix D is diagonal and gives the diffusion coefficient for each species.

The fact that it is diagonal means that the different substances don't change into each other.

One-Components Systems

Simple diffusion $R(q) = 0$:

$$\frac{\partial q}{\partial t} = D \frac{\partial^2 q}{\partial x^2}$$

We already derived this:

- ▶ arises from Gaussian random walk
- ▶ the change at x over time is the stuff that diffuses out to the left and right minus the stuff that diffuses in from the left and right

```
1 q = zeros(1000)
2 q[400:600] = 1.0
3 result = []
4 for i in range(20000):
5     if i%20==0: result.append(q.copy())
6     q += 0.2*laplacian(q)
7 subplot(121); imshow(array(result))
8 subplot(122); ylim((-0.1,1.1)); plot(result[0]); plot(result[500]);
    plot(result[-1])
```

Fisher's equation:

$$\frac{\partial q}{\partial t} = D \frac{\partial^2 q}{\partial x^2} + q(1 - q)$$

Motivation:

- ▶ animal populations
- ▶ diffusion term: spatial “diffusion” / migration of individuals
- ▶ logistic term: growth of advantageous allele over original wild type

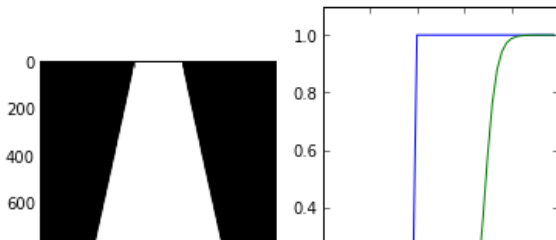
Behavior:

- ▶ traveling wave solutions

1D Fisher equation - traveling wavefront

```
1 q = zeros(1000)
2 q[400:600] = 1.0
3 result = []
4 for i in range(20000):
5     if i%20==0: result.append(q.copy())
6     q += 0.01*(laplacian(q)+q*(1-q))
7 result = array(result); print amin(result),amax(result)
8 subplot(121); imshow(array(result))
9 subplot(122); ylim((-0.1,1.1)); diffusionplot(result[0][380:430]);
    plot(result[-1][150:200])
```

0.0 1.0



2D Fisher equation - traveling wavefront

```
1 q = zeros((w,h))
2 q[w//2:w//2+10,h//2:h//2+10] = 1N = 100
3 for trial in range(10000):
4     u = 0.5*(1+0.1*randn(N))
5     v = 0.1*(1+0.1*randn(N))
6     a = exprand(0.001,0.1)
7     b = exprand(0.001,0.1)
8     r = exprand(0.001,0.1)
9     for t in range(10000):
10        u += 0.02*laplacian(u)
11        v += 0.04*laplacian(v)
12        u -= r*u*v**2
13        v += r*u*v**2
14        u += a*(1-u)
15        v -= b*v
16    delta = amax(u)-amin(u)
17    if delta>0.9:
18        print a,b,r,delta
19        break
20 q[w//2+30:w//2+40,h//2:h//2+10] = 1
21 def rd():
22     global q
23     for i in range(50000):
24         yield q
```

Turing Systems

Two components:

Let's consider two components now. This turns the reaction-diffusion equation into:

$$\begin{pmatrix} \partial_t u \\ \partial_t v \end{pmatrix} = \begin{pmatrix} D_u & 0 \\ 0 & D_v \end{pmatrix} \begin{pmatrix} \partial_{xx} u \\ \partial_{xx} v \end{pmatrix} + \begin{pmatrix} F(u, v) \\ G(u, v) \end{pmatrix}$$

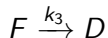
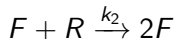
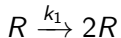
Turing systems:

- ▶ two morphogens
- ▶ first morphogen is autocatalytic and diffuses slower
- ▶ second morphogen is inhibitory and diffuses faster

Similar to the three component system we discussed before, but local growth, slow and fast diffusion are mixed up.

Reaction Kinetics:

Chemical reactions:



Differential equations:

$$[\dot{R}] = k_1[R] - k_2[F][R]$$

$$[\dot{F}] = k_2[F][R] - k_3F$$

Gray-Scott Systems:

Autocatalytic reaction: $U + 2V \rightarrow 3V$

Corresponding rates: $[\dot{U}] = -[U][V]^2$

Replenishment: $[\dot{U}] = f(1 - [U])$

Destruction: $[\dot{V}] = -(f + k)[V]$

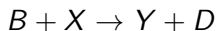
Equations:

$$\dot{u} = r_u \nabla^2 u - uv^2 + f(1 - u)$$

$$\dot{v} = r_v \nabla^2 v + uv^2 - (f + k)v$$

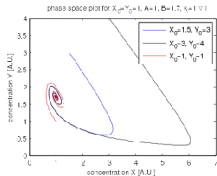
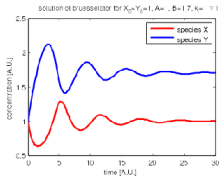
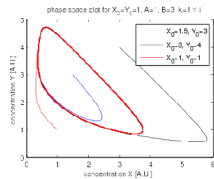
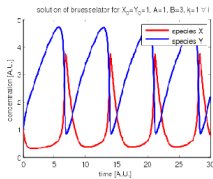
(Examples: later)

Brusselator:



(Note that concentrations of A , B , C , and D determine rate constants.)

<http://iupr1.cs.uni-kl.de/~tmb/simso-videos/brusselator.mp4>



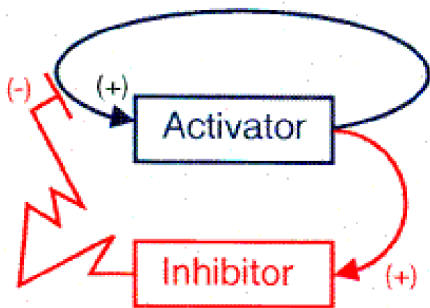
Gierer-Meinhardt Model:

$$\dot{a} = \alpha a^2 - \mu_a a + D_a \nabla^2 a + \rho_a$$

$$\dot{h} = \beta a^2 - \mu_h h + D_h \nabla^2 h + \rho_h$$

<http://iupr1.cs.uni-kl.de/tmb/simso-videos/gierer-meinhard-spots.mp4>

<http://iupr1.cs.uni-kl.de/tmb/simso-videos/gierer-meinhardt-labyrinth.mp4>



FitzHugh-Nagumo Equation with Diffusion:

We already discussed this as a model of the action potential.

$$\dot{u} = D_u \nabla^2 u + f(u) - sv$$

$$\tau \dot{v} = D_v \nabla^2 v + u - v$$

With:

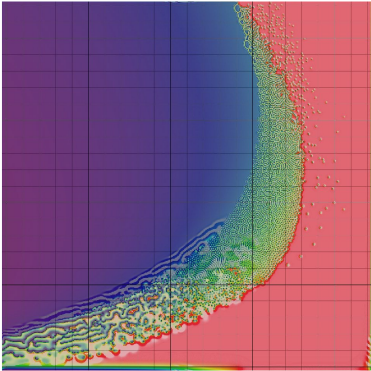
$$f(u) = u^3 - \kappa$$

Here, the “activator” is the sodium channels, and the “inhibitor” is the potassium channels.

<http://iupr1.cs.uni-kl.de/~tmb/simso-videos/fitzhugh-tip-splitting.mp4>

Exploring Gray Scott

Finding parameters that work can be tricky. A simple method that often works is to run the equations on a variable parameter space.

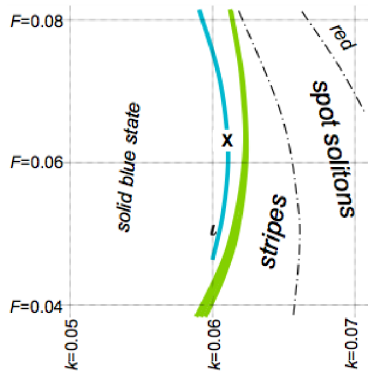


You can explore this space here:

<http://mrob.com/pub/comp/xmorphia/formula>

<http://iupr1.cs.uni-kl.de/tmb/simso-videos/Gray-Scott-F300-k630-fr47.mp4>

<http://iupr1.cs.uni-kl.de/tmb/simso-videos/Gray-Scott-F380-k630-fr64.mp4>



U-Skate World:

<http://iupr1.cs.uni-kl.de/tmb/simso-videos/uskate-world-complexity.mp4>